

## Piezoelectric properties of lead-free $(1-x)(\text{Na}_{0.5}\text{K}_{0.5})\text{NbO}_3\text{-xLi}(\text{Sb}_{0.17}\text{Ta}_{0.83})\text{O}_3$ ceramics

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The  $(1-x)(\text{Na}_{0.44}\text{K}_{0.52})\text{Nb}_{0.84}\text{O}_3\text{-xLi}_{0.04}(\text{Sb}_{0.06}\text{Ta}_{0.1})\text{O}_3$  ( $0.01 \leq x \leq 0.05$ ) lead free ceramics were prepared using a conventional mixed oxide method. Adding various contents of  $\text{Li}(\text{Sb},\text{Ta})\text{O}_3$  to  $(\text{Na},\text{K})\text{NbO}_3$  makes it easy to conduct the sintering processes of pure  $(\text{Na},\text{K})\text{NbO}_3$  which is difficult to sinter. It was found that the piezoelectric properties of  $(1-x)\text{NKN-xLST}$  ( $x = 0.02$ ) ceramic sintered at 1080 °C for 2 h has a piezoelectric constant and a planar electromechanical coupling coefficient of 159 pC/N and 38% respectively. It is found that the addition of LST can improve poling procedures of the ceramics because of substitution. These solid solution ceramics look attractive candidates for lead-free piezoelectric applications.

**Key words:** Lead-free, NKN-LST, Piezoelectric constant, Dielectric constant.

### Introduction

The morphotropic phase boundary (MPB) in lead-based materials such as  $\text{Pb}(\text{Zr}_{1-x}\text{Ti}_x)\text{O}_3$  (PZT),  $(1-x)\text{Pb}(\text{Mg}_{1/3}\text{Nb}_{2/3})\text{O}_3\text{-xPbTiO}_3$  (PMN-PT) and  $(1-x)\text{Pb}(\text{Zn}_{1/3}\text{Nb}_{2/3})\text{O}_3\text{-xPbTiO}_3$  (PZN-PT), plays a special role [1-3]. It is believed that a sample with its composition around the MPB shows excellent piezoelectric and dielectric properties. However, lead-based materials are dominated by containing more than 60 wt% lead [4]. The lead evaporation during the sintering process of PZT and discarded PZT components pollute the environment and are harmful to health [5-6]. Therefore, it is an urgent task that environment-friendly piezoelectric materials are developed to replace lead-based materials [7]. Among them,  $(\text{Na},\text{K})\text{NbO}_3$  (hereafter NKN) exhibits an MPB at around 50% Na and 50% K separating two orthorhombic phases and as for PZT, an increase in piezoelectric and dielectric properties for compositions close to the MPB [9-13]. Hot-pressed NKN ceramics have been reported to possess a high Curie temperature ( $T_c \sim 420$  °C), a large piezoelectric longitudinal response ( $d_{33} \sim 160$  pC/N) and a high planar coupling coefficient ( $k_p \sim 45\%$ ) [14-15]. However, for NKN ceramics there is the need for special processing of the starting materials, a sensitivity of properties to nonstoichiometry and a complex densification process [16-17]. Sintering under ordinary conditions shows lower properties ( $d_{33} \sim 80$  pC/N) due to a poor density. Thus, adding other perovskite compounds to form solid solutions with NKN has been performed to obtain lead-free materials comparable with lead-based materials [18]. The goal of this paper is to obtain a system of NKN-based

piezoelectric materials with enhanced piezoelectric, dielectric properties by adding different amounts of  $\text{Li}(\text{Sb},\text{Ta})\text{O}_3$  (hereafter LST).

### Experimental

The chemical molecular formulae used in this experiment for the perovskite ceramics with (Na, K, Li) complex A-sites and (Nb, Sb, Ta) complex B-sites is  $(1-x)\text{NKN-xLST}$  ceramics ( $0.01 \times 0.05$ ). For specimens prepared by the conventional mixed oxide method from  $\text{Na}_2\text{CO}_3$ ,  $\text{K}_2\text{CO}_3$ ,  $\text{Nb}_2\text{O}_5$ ,  $\text{Li}_2\text{CO}_3$ ,  $\text{Sb}_2\text{O}_5$ ,  $\text{Ta}_2\text{O}_5$  as the starting materials, these powders were separately dried in an oven at 100 °C for 4 h. They were ball-milled for 24 h using  $\text{ZrO}_2$  balls in alcohol. After drying at 110 °C for 24 h, the powders were calcined at 850 °C for 5 h. The calcined powders were pressed into disk samples of φ 12 mm. The samples were sintered at 1080 °C for 2 h. After the samples were polished to 1.0 mm thickness, silver paste was screen-printed on the surfaces as electrodes and then fired at 400 °C for 10 minutes. We used X-ray diffraction (XRD) to analyze the crystallinity and microstructures. The dielectric properties were measured using an LCR meter (PM6306, Pluke). Hysteresis loops of the samples were measured by a Sawyer-Tower circuit. The samples were poled under a DC field of 4 kV/mm for 20 minutes. The piezoelectric strain constant  $d_{33}$  was measured by a  $d_{33}$  meter (Channel Product DT-3300). The electromechanical coupling factor  $k_p$  was calculated by measuring the anti-resonance and resonance frequencies. The relative density of the sintered samples was measured by the Archimedes method.

### Results and Discussion

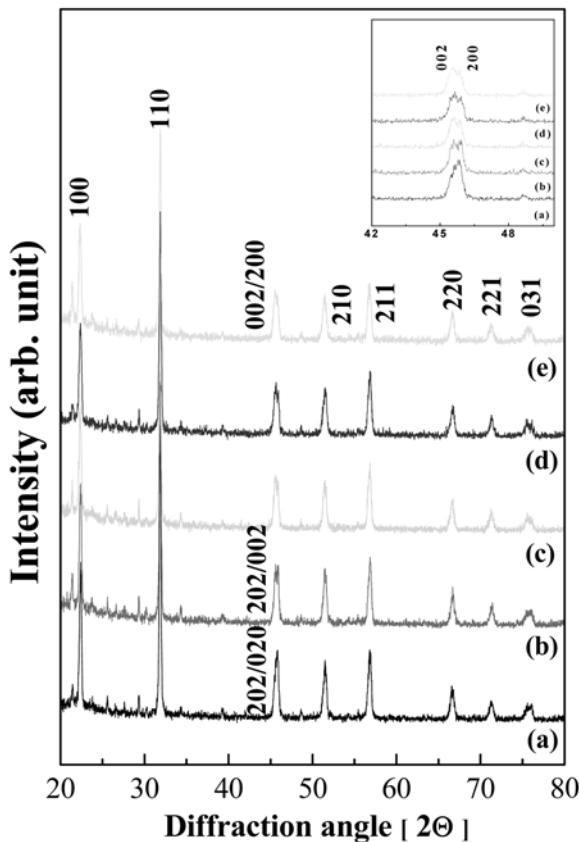
Fig. 1 shows the X-ray diffraction patterns of  $(1-x)$

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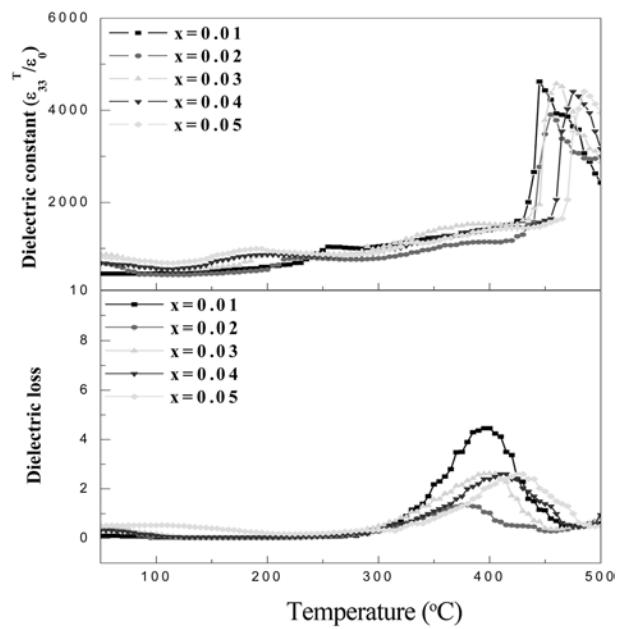
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**Fig. 1.** XRD patterns of the  $(1-x)NKN-xLST$  ( $x$  mol%) ceramics (a) 0.01 mol% (b) 0.02 mol% (c) 0.03 mol% (d) 0.04 mol% (e) 0.05 mol%.

$(Na_{0.44}K_{0.52})Nb_{0.84}O_3-xLi_{0.04}(Sb_{0.06}Ta_{0.1})O_3$  ( $0.01 \leq x \leq 0.05$ ) ceramics sintered at  $1080^\circ C$  for 2 h. The pure NKN ceramics are usually sintered at  $1150^\circ C$ . For  $(1-x)NKN-xLST$  ceramics not only is suppressed the sintering temperature lowered but also volatilization the of  $Na^+$ ,  $K^+$ . As seen from these patterns, all samples show a pure perovskite phase, and no secondary phase could be identified. This indicates that LST has diffused into the NKN lattice and formed solid solutions. In this case,  $Li^{+}$  ion diffuses to the  $Na^+$ ,  $K^+$  sites of NKN, while  $Sb^{5+}$ ,  $Ta^{5+}$  occupies the  $Nb^{5+}$  sites of NKN. The inset of Fig. 1 exhibits the variation of peak transitions of  $(1-x)NKN-xLST$  ceramics near  $2\theta = 45^\circ$ . The orthorhombic phase is characterized by  $(202)/(020)$  peak splitting at about  $45^\circ$ . Also, the tetragonal phase is characterized by  $(002)/(200)$  peak splitting at about  $45^\circ$ . Therefore, at room temperature, it can be concluded that  $(1-x)NKN-xLST$  ceramics have orthorhombic symmetry at  $x < 0.02$ , has a tetragonal structure at  $x > 0.03$  and has a coexistence of orthorhombic and tetragonal phases at  $x = 0.02, 0.03$ . As a result of these phenomena, we expect before further experiments, that there is a composition of the so-called MPB in  $(1-x)NKN-xLST$  ceramics at  $x = 0.02, 0.03$ .

The dielectric properties versus temperature measurements of unpoled  $(1-x)NKN-xLST$  ( $0.1 \leq x \leq 0.5$ ) ceramics at

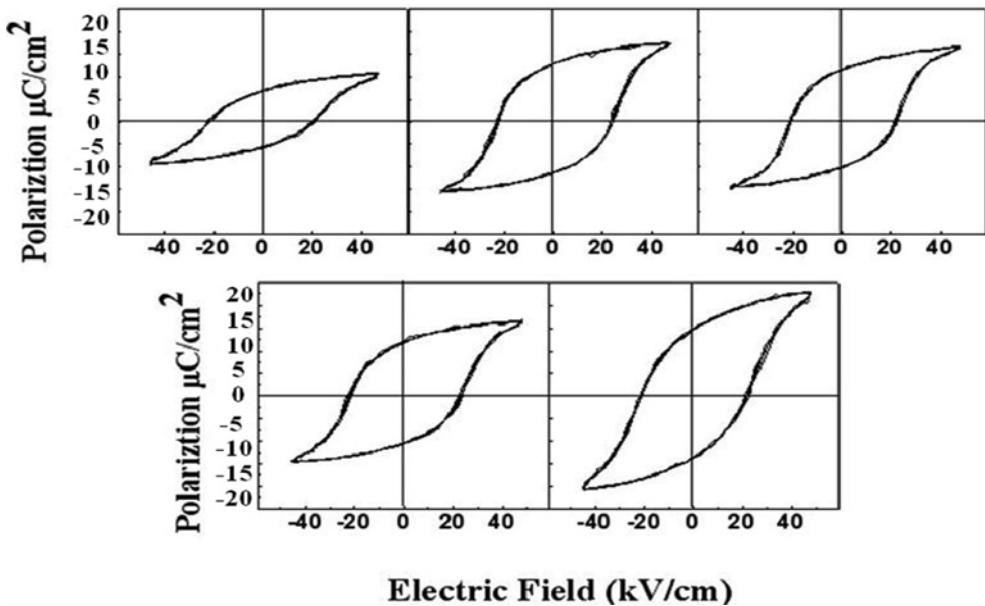


**Fig. 2.** The dielectric constant and loss of the  $(1-x)NKN-xLST$  ( $x$  mol%) ceramics at frequency of 1 kHz.

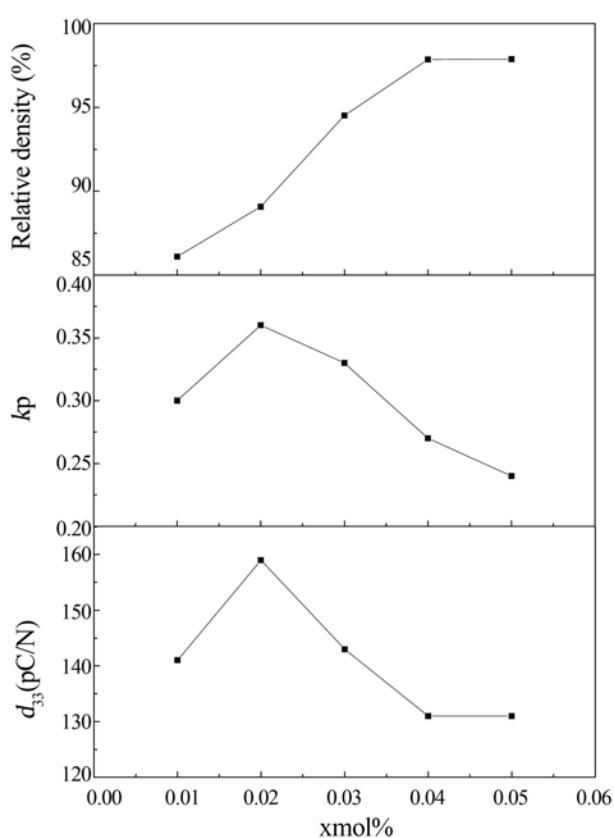
1 kHz are shown in Fig. 2. For the pure NKN ceramic, two phase transition temperatures are reported at approximately  $450^\circ C$  and  $250^\circ C$ , conforming to the phase transition of tetragonal-cubic ( $T_c$ ) and orthorhombic-tetragonal ( $T_{O-T}$ ) respectively. The samples with  $x = 0.01$ , show similar behavior to the pure NKN ceramics with respect to NKN, the  $T_{O-T}$  is shifted to a lower temperature as the LST content  $x$  is increased. It can be inferred that the lithium and antimony additions stabilize the tetragonal structure. However, the  $T_c$  shifts to a higher temperature with an increase in the LST content from 0.01 to 0.05. In the case of the lithium and tantalum dominantly substituted  $(1-x)NKN-xLST$  ceramics,  $T_c$  is increased by increased substitution.

Fig. 3 shows the polarization versus electric field (P-E) hysteresis loops for the  $(1-x)NKN-xLST$  sintered  $1080^\circ C$  for 2 h. As shown in the figures, the polarization loops grow wider with an increase in the LST content. The remnant polarization of the 0.98NKN-0.02LST is about  $11.3 \mu C/cm^2$ , which increased with the content of LST, exhibiting a maximum value of  $14.2 \mu C/cm^2$ . When the LST content was 0.05. The increase in  $P_r$  could be related to an increase in the bulk density. Therefore, it is considered that because the radius of  $Li^+$  ( $0.68 \text{ \AA}$ ),  $Ta^{5+}$  ( $0.70 \text{ \AA}$ ),  $Sb^{5+}$  ( $0.62 \text{ \AA}$ ) ions are similar to that of  $Nb^{5+}$  ( $0.70 \text{ \AA}$ ),  $Li^+$ ,  $Ta^{5+}$  and  $Sb^{5+}$  ion these entered the  $Nb^{5+}$  sites of the  $(1-x)NKN-xLST$  ceramics, and might behave as a hardener.

The compositional dependence of poled piezoelectric properties and relative density were measured for the NKN-LST system, as shown in Fig. 4. Similar to PZT, it is well known that the electrical properties rapidly increase at the MPB. For  $(1-x)NKN-xLST$  ceramics, the electromechanical coupling factor  $k_p$  varied in the range of 24%-38%



**Fig. 3.** Hysteresis loops of the (1-x)NKN-xLST (x mol%) ceramics (a) 0.01 mol%, (b) 0.02 mol% (c) 0.03 mol% (d) 0.04 mol% (e) 0.05 mol%.



**Fig. 4** Piezoelectric properties and relative density of the (1-x)NKN-xLST (x mol%) ceramics.

and the piezoelectric constant  $d_{33}$  was in the range of  $131\text{ pC/N}$ - $159\text{ pC/N}$ , which had enhanced values as compared with the pure NKN ceramic. Poled samples show the

highest values of the electromechanical coupling coefficient  $k_p = 38\%$  and piezoelectric coefficients  $d_{33} = 159 \text{ pC/N}$  for 0.98NKN-0.02LST ceramics. The improvement of piezoelectric properties could be explained by an increase in the bulk density. With an increase in the LST content the piezoelectric properties increased. However, for LST contents in excess of 0.02, the piezoelectric properties decreased due to a transformation of the microstructure. It is significant that the MPB plays the most important factor in improving piezoelectric properties of NKN-LST ceramics. Also, Fig. 4 shows the effect of the substitutional content. All samples reached a density of more than 85% of the theoretical density at a sintering temperature of  $1080^\circ\text{C}$ . The relative density increased with the LST content. However, with an increase in the LST content above 0.02 mol%, the piezoelectric properties of (1-x)NKN-xLST ceramics were decreased due to the excessive substitution by LST.

## Conclusion

(1-x)(Na<sub>0.44</sub>K<sub>0.52</sub>)Nb<sub>0.84</sub>O<sub>3</sub>-xLi<sub>0.04</sub>(Sb<sub>0.06</sub>Ta<sub>0.1</sub>)O<sub>3</sub> ( $0.01 \leq x \leq 0.05$ ) lead free ceramics were prepared using a conventional mixed oxide method. From the X-ray diffraction analysis, we found that (1-x)NKN-xLST ceramics were crystallized and showed a MPB (orthorhombic and tetragonal structures) at  $x = 0.02, 0.03$ . The P-E hysteresis loops for the 0.98NKN-0.02LST ceramics sintered at  $1080^\circ\text{C}$  shows a high remnant polarization of  $1 \mu\text{C}/\text{cm}^2$ . Also, we found that 0.98NKN-0.02LST ceramics sintered at  $1080^\circ\text{C}$  have a relatively high piezoelectric coefficient ( $d_{33}$ ) of  $159 \text{ pC/N}$  and electromechanical coupling coefficient  $k_p = 38\%$ . These excellent piezoelectric properties mean that this lead-free 0.98NKN-0.02LST piezoelectric ceramic can be a promising lead-free piezoelectric material.

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## References

1. Ruzhong Zuo, Jurgen Rodel Renzheng Cen and Longtu Li, J. Am. Ceram. Soc., 89[6] (2006) 2010-2015.
2. S.-H. Park, C.-W. Ahn, S. Nahm and Jae-Sung Song, The Japan Society of Applied Physics, 43 (2004) L120072-L1074.
3. Y.-H. Kim, D.-Y. Heo, W.-P. Tai and J.-S. Lee, Journal of the Korean Ceramic Society, 45[6] (2008) 363-367.
4. S.-J. Park, H.-Y. Park, K.-H. Cho, S. Nahm and H.-G. Lee, Materials Research Bulletin 42 (2008) 3580-3586.
5. Yasuyoshi Saito, Hisaaki Takao, Toshihiko Tani, Tatsuhiko Nonoyama, Kazumasa Takatori, Takahiko Homma, Toshiatsu Nagaya, and Masaya Nakamura, Nature, 432 (2004) 84-87.
6. Z.-X. Chen, Y. Chen and Y.-S. Jiang, J. Phys. Chem. B, 106 (2002) 9986-9992.
7. S.-H. Moon, Y.-S. Ham, Y.-H. Lee, S.-M. Nam and J.-H. Koh, Journal of the Korean Physical Society, 56[1] January (2010) 399-403.
8. Ruzhong Zuo and Chun Ye, Appl. Phys. Lett. 91 (2007) 062916.
9. M. Jiang, X. Li, J. Liu, J. Zhu, X. Zhu, L. Li, Q. Chen, J. Zhu and Dingquan Xiao, J. Alloys Compounds 479 (2009) L18-L21.
10. I.J. Cho, K.S. Yun and H.J. Nam, J. Electrical Engineering & Technology, 6[1] (2011) 119-122.
11. M.S. Kim, Y.M. Jeon, Y.M. IM, Y.H. Lee and T.H. Nam, Trans. Electr. Electron. Mater., 12[1] (2011) 20-23.
12. H.J. Bae, J.K and J.P. Hong, J. Electrical Engineering & Technology, 1[1] (2006) 120-126.
13. J.H. Heo, Trans. Electr. Electron. Mater., 11[6] (2010) 275-278.
14. X.K. Niu, J.L. Zhang, L. Wu, P. Zheng and T.R. Shrout, Appl. Phys. Lett. 91 (2007) 132913.
15. S.P. Nam, S.G Lee, S.G. Bae and Y.H. Lee, J. Electrical Engineering & Technology, 2[1] (2007) 98-101.
16. L. Wu, D.Q. Xiao, J.G Wu, Y. Sun, D.M. Lin, J.G Zhu, P. Yu, Y. Zhuang and Q. Wei, J. Eur. Ceram. Soc. 28 (2008) 2963-2968.
17. Y.F. Chang, Z.P. Yang, L.R. Xiong, Z.H. Liu and Z.L. Wang, J. Am. Ceram. Soc. 91 (2008) 2211-2216.
18. S.H. Lee, J. Electrical Engineering & Technology, 2[1] (2007) 102-105.